(05/16-1)

5252

B. Sc. EXAMINATION

(Sixth Semester)

CHEMISTRY (INORGANIC)

Paper XVIII (CH-304)

ime: Three Hours Maximum Marks: 27

iote: Attempt Five questions in all. Question number 1 (7 questions of 1 mark each) is compulsory. In addition to question no. 1, candidate are required to attempt two questions from each Section A and B.

- (a) What is organometallic compound?
- (b) Give the IUPAC names of $Fe(CO)_5$ and $Fe_2(CO)_4(C_5H_5)_2$.
- (c) What is ionisation constant?
- (d) Why is ammonia termed as a base through it does no containt OH ions?

54) B-5252 P.T.O.

(e) What are nitrogenases?

- (f) What is the role of iodine in human body?
- (g) What are silicon fluids? $1\times7=7$

Section A

- (a) What is Zeise salt and how is it prepared? Discuss the nature of bond in this compound.
 - (b) How does Wilkinson's catalyst play its role in hydrogenation of alkenes? 2

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- (a) Define acids and bases in term of Luxflood concept. Give its advantages.
 - (b) Justify by Lewis concept that CO₂ is an acid.2
- 4. (a) Which factros are responsible for kinetic instability of transition metal sigma bonded organometallic compounds? 3
 - (b) Describe the application of HSAB principle.

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Section B

What are metalloporphyrins? Explain the structure of hemoglobin and mycglobin.

- (a) Discuss biological importance of Ca²⁺?
 How does it differ from the Mg²⁺?
 - (b) Discuss important properties of Silicon.

(a) Discuss the structure of $(NPCl_2)$ and $(NPCl_2)_3$.

- (b) Write brief notes on the following: 2
 - (i) Silicon resins
 - (ii) Silicon rubber.

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